

A Review on Eco-Friendly Synthesis of BiVO₄ Nanoparticle and its Eclectic Applications

Suresh Ghotekar^{1,2,✉}, Khanderao Pagar³, Shreyas Pansambal¹, H.C. Ananda Murthy⁴,
Rajeshwari Oza^{1,✉}



Received: September 12, 2020 / Accepted: September 15, 2020 / Published Online: October 13, 2020

ABSTRACT. The paper presents a review of green syntheses and selective applications of bismuth vanadate nanoparticles (BiVO₄ NPs). Generally, ample number of biomolecules exists in plant extracts and these are mainly accountable for the facile green synthesis of BiVO₄ NPs. Moreover, BiVO₄ NPs has been widely researched in chemistry, biotechnology, physics and biochemistry fields due to their interesting technological chemical, biological, ionic conductivity and ferro-elastic properties. It can also be used in diverse fields, such as sensors, photocatalysis, water splitting and antimicrobial activity. Till date, BiVO₄ NPs has been synthesized by various known physical and chemical approaches. The article mainly discusses the green synthesis of BiVO₄ NPs via plant extracts. Moreover, this article shows a detailed overview of the green synthesis, characterization and significant applications of BiVO₄ NPs.

Keywords: BiVO₄ nanoparticle; Applications; Green nanotechnology; Plant extracts.

INTRODUCTION

Nanotechnology is a multidisciplinary branch of modern research comprising of fabrication methodology and manipulation of particle's size distribution with selective morphology of functional nanomaterials. The manipulation of their biological, chemical, electrical, optical or physical properties can generate functional nanomaterials with size ranging

from 1 to 100 nm. Modifications of these functional materials can create novel materials with enhanced or improved features/properties.¹⁻² Functional nanomaterials have stupendous applications of areas ranging from environment, biomedical and health care, food preservation, cosmetics, fuel cells, water purification, drug delivery and gene delivery, defense, chemical industries, space industries, ceramics, electronics, energy, sensors, single electron transistors, textiles, agricultures, solar cells, catalysis, light emitters, fuel, and antimicrobial.¹⁻²⁵ This is due to progress of reactivity when compared to their bulk counterparts and/or micro-sized since functional nano-scaled materials evince larger surface area to volume ratio.¹ Hence, there is a massive interest in the production of NPs for researchers working in the discipline of nanoscience and nanotechnology.

There are two significant approaches to obtain nanomaterials, they are called the top-down and bottom-up approaches.¹⁻⁵ Top-down method incorporates breaking down bulk material into tiny particles through size reduction using different approaches such as, electric arc, grinding, sputtering, ball milling, and thermal ablation. Bottom-up approaches is basically synthesizing NPs from smaller entities such as joining atoms, molecules and tiny particles. The bottom up approach mostly relies upon

✉ Corresponding author.

E-mail addresses: ghotekarsuresh7@gmail.com (S. Ghotekar),
rajeshwarikaraswat@gmail.com (R. Oza)

¹ Department of Chemistry, S.N. Arts, D.J.M. Commerce and B.N.S. Science College, Sangamner 422 605, Savitribai Phule Pune University, Maharashtra, India

² Department of Chemistry, Sanjivani Arts, Commerce and Science College, Kopergaon 423 603, Savitribai Phule Pune University, Maharashtra, India

³ Department of Chemistry, S.S.R. College of Arts, Commerce and Science College, Silvassa 396 230, Savitribai Phule Pune University, Dadra and Nagar Haveli, India

⁴ Department of Applied Chemistry, School of Applied Natural Sciences, Adama Science and Technology University, P.O. Box: 1888, Adama, Ethiopia

chemical and biological approaches. This methodology has an immense advantage which is the enhanced possibility of syntheses of NPs with minimum efforts.² In particular, BiVO_4 is a non-toxic yellow pigment and n-type semiconductor possesses diverse interesting technological properties (Fig. 1), such as ferro elasticity, photostability and ionic conductivity.²⁶

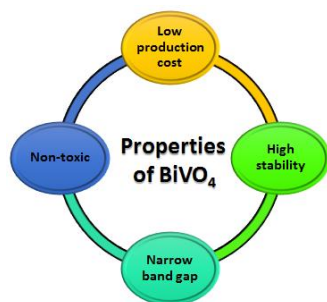


Fig. 1: Properties/features of BiVO_4 NPs.

Hence, BiVO_4 has attracted noteworthy interest due to its outstanding features, such as low band gap, resistance to corrosion, good dispersibility, non-toxicity and excellent photocatalytic result in organic pollutant degradation under visible-light.²⁷ Moreover, its adaptable optical and electronic properties with a band gap ~ 2.4 eV makes it a significant candidate for harvesting solar light.²⁸⁻²⁹ It is notable that BiVO_4 mainly exists in three crystalline phases such as, tetragonal zircon phase (t-z), monoclinic scheelite phase (m-s) and tetragonal scheelite phase (t-s).²⁷ Heretofore, BiVO_4 NPs can be easily synthesized using multifarious methods such as drop casting method,³⁰ solution combustion method,³¹ microwave-irradiation,³² hydrothermal,³³ co-precipitation method,³⁴ template free approach,³⁵ flame-assisted,³⁶ thermal decomposition,³⁷ low temperature method,³⁸ ultrasonication,³⁹ mechanical milling,⁴⁰ polymer assisted co-precipitation,⁴¹ mechanochemical,⁴² microemulsion,⁴³ ball milling,⁴⁴ molten salt method,⁴⁵ solvothermal,⁴⁶ reflux method,⁴⁷ sol gel,⁴⁸ sonochemical,⁴⁹ and metal organic decomposition.⁵⁰ Unfortunately, it often requires complicated and rigorous preparation conditions, time and energy consumption in the conventional synthesis methods. Therefore, safer, straightforward, biocompatible and economic process of uniform size distribution of BiVO_4 NPs still has lot of challenges.

Here, we covered the current scenario of research on the eco-friendly synthesis of BiVO_4 NPs with their benefits over known conventional routes. The main goal of this

literature survey is to provide details of plants those are reported on biogenic syntheses and their eclectic applications. Overall, our objective is to critically describe green synthesis protocols and versatile applications of NPs that will be advantageous to researchers involved in this emerging sector. Thus, this perspective review article intends to present reports on green syntheses, characterization techniques and eclectic applications of the BiVO_4 NPs using plants extract.

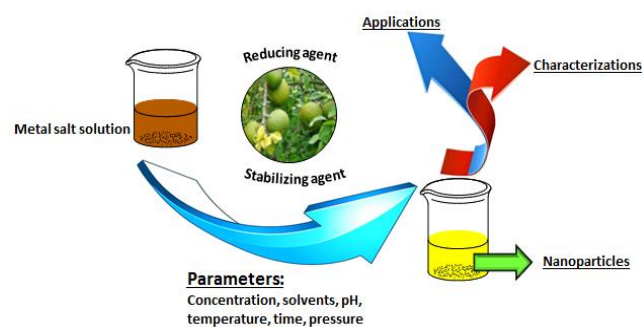


Fig. 2: Process of green synthesis of NPs.

GREEN SYNTHESIS OF METAL OXIDE NPS

Nowadays, green syntheses of functional nanomaterial imply the creating of NPs and/or nanomaterials without using noxious chemicals that generates pernicious by-products. In other words, green methodology is an eco-benevolent protocol to produce NPs where it is not ruinous to the ecosystem and human health. It is true facts that known conventional methodology can produce NPs in large quantities of desired shapes and size. Notwithstanding, these methodologies require high cost, complicated, tedious, noxious and outdated protocols.¹⁻¹⁰ In contrast to the conventional protocols, a green approach has immense advantages such as swift, simple manufacturing protocol, sustainable, straightforward and economically affordable. From the viewpoint of the ecosystem, green approaches for synthesizing NPs are considered, as a specific chemical it is not required to be reduced and stabilized, and furthermore its fabrication can be done under mild conditions.² In the green synthesis of NPs, raw materials, plants extracts (fruits, flowers, seeds, leaf, roots, etc.), enzymes, microbes, and fungi are utilized to prepare functional NPs.³ The control of the size and morphology of green synthesized NPs and their precise mechanism of formation is still two significant challenges in nanobiotechnology.¹⁸ In general, three routes of green syntheses using plant extracts (Fig. 2),

green synthesis using microorganisms, and low-temperature synthesis have been utilized for synthesis functional NPs.

GREEN SYNTHESIS OF BiVO₄ NPS FROM PLANTS EXTRACT

There are very scanty reports on green syntheses of BiVO₄ NPs. However, employing plant extracts, natural reduction and stabilization of bismuth metal and vanadium metal into BiVO₄ NPs are the simplest, swift, inexpensive, sustainable and environmentally gracious procedures in green chemistry.⁵¹ Green synthesis by plant extracts has immense benefits, including scalability, biocompatibility, and medical applicability.⁴ In the synthesis of functional NPs employing selected plant extracts, the plant extract is simply mixed with the metal salt solution at proper temperature and the reaction is complete in a few minutes. The metal reduction is ascribed to the various phytoconstituents which are available in the plant extract such as tannins, polysaccharides, saponins, terpenoids, proteins, phenols and flavonoids.³ So far, several plant extracts have been used in the synthesis of BiVO₄ NPs. Mohamed et al.⁵¹ reported the green synthesis of rod shaped (Fig. 3.) BiVO₄ NPs by reducing bismuth nitrate and vanadium sulphate with the help of *Callistemon viminalis* flower extract as a reducing and stabilizing agent with an average crystallite size of BiVO₄ NPs in the range of 7.34 nm. At their experience, pH was kept among 3.5 and the temperature was at 100°C during the experiment. Same author has described two more reports on green synthesis of BiVO₄ NPs using *Callistemon viminalis* flower extract.^{53, 55}

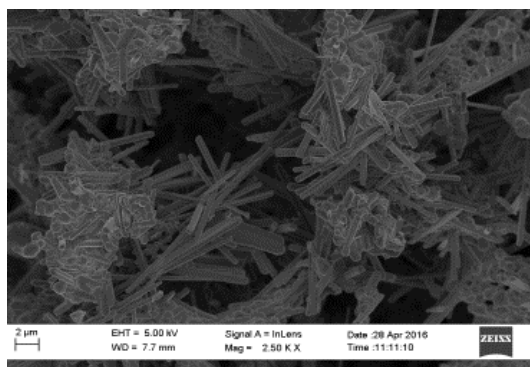


Fig. 3: TEM image of BiVO₄ NPs.⁵³

Manjunatha et al.⁵² used an aqueous fruit extract of *Citrus lemon* to produce BiVO₄ NPs that had an average particle size of 75 nm with a bang gap energy ranging

from 2.6 to 2.8 eV. Phytosynthesis of BiVO₄ nanorods using a fruit extract from *Hyphaene thebaica* was reported by Khalil et al.⁵⁴ with an average size of 7 nm. In this research, the mixture was stirred 1 h at 100°C. The brief protocol of phytosynthesis of BiVO₄ NPs by *Hyphaene thebaica* fruit extract is described in Fig. 4.

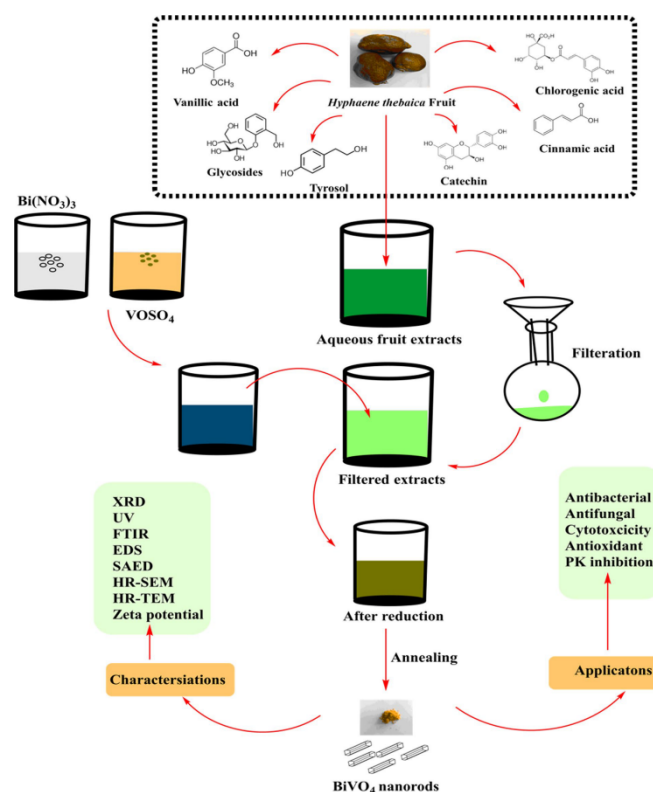


Fig. 4: Phytosynthesis of BiVO₄ NPs using *Hyphaene thebaica* fruit extract.⁵⁴

In another study, Pramila et al.⁵⁶ reported the green synthesis of BiVO₄ NPs using *Aegle marmelos* fruit juice as a fuel. Green synthesis of BiVO₄ NPs using few plants extracts and their characterization techniques are summarized in Table 1.

RECENT APPLICATIONS OF BIO-SYNTHEZED BiVO₄ NPS

BiVO₄ NPs have eclectic applications relying upon the diverse properties they manifest, which are significantly affected by their morphology, size, and optical traits. Thus, the synthesis approaches being imperative parameters for controlling all these properly. Some of these applications include sensors, photocatalysis, water splitting and antimicrobial activity etc. We have portrayed their advantageous applications to emphasize their momentous outcomes as direction to new researchers for future prospects. Mohamed et al.⁵¹ synthesized rod shaped BiVO₄ NPs using *Callistemon*

viminalis and reported their photocatalytic activity. They demonstrated that, the photocatalytic degradation of methylene blue (MB) dye up to 82% in 5 h under solar visible at room temperature by using BiVO₄ NPs as an efficient photocatalyst. The schematic mechanism of MB dye degradation is depicted in Fig. 5.

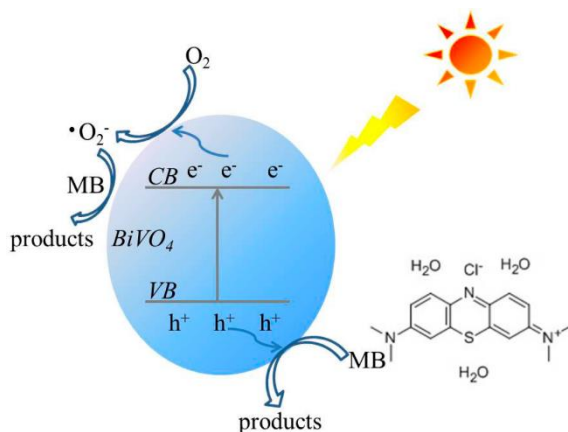


Fig. 5: Photocatalytic performance of BiVO₄ NPs using MB dye.⁵⁷

Manjunatha et al.⁵² reported the facile green synthesis of BiVO₄ NPs using Citrus lemon juice extracts and studied their photocatalytic and electrochemical activities. The synthesized BiVO₄ NPs catalyzed up to 90.6% degradation of *Indigo Carmine* dye within 150 min. Herewith, they also examined the electrochemical

behavior of BiVO₄ modified electrode for the efficient detection of Hg (II) using electrochemical protocol. Khalil et al.⁵⁴ reported the phytosynthesis of BiVO₄ NPs using *Hyphaene thebaica* fruits extract and also described their antimicrobial, protein kinase (PK) inhibition, antioxidant, hemolysis and antiviral activities. These synthesized BiVO₄ NPs showed excellent antimicrobial performance against *Staphylococcus epidermidis* (ATCC 14490), *Bacillus subtilis* (ATCC 6633), *Klebsiella pneumonia* (ATCC 13883), *Escherichia coli* (ATCC 15224) and *Pseudomonas aeruginosa* (ATCC 9721), *Aspergillus flavus* (FCBP 0064), *Aspergillus fumigatus* (FCBP 66), *Mucor sp.* (FCBP 300), *Aspergillus niger* (FCBP 0918) and *Fusarium solani* (FCBP 434) using simple well diffusion assay. In addition, significant and potent protein kinase inhibition is reported for BiVO₄ nanorods, which suggest their potential anticancer properties. Besides, phytosynthesized BiVO₄ nanorods exhibited good antioxidant activity using 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging potential and total antioxidant capacity protocols. Furthermore, antiviral activity of BiVO₄ nanorods was carried out against human rhabdomyosarcoma cells (RD), human laryngeal carcinoma (HEp-2 cells) and L20B cells (mouse fibroblast cells).

Table 1: Green synthesis of BiVO₄ NPs using plant extracts with size and shape.

Name of Plants	Part	Characterization techniques	Shape	Size (nm)	Ref.
<i>Callistemon viminalis</i>	Flowers	XRD, UV-Vis, HRTEM, SEM, EDS, FTIR, XPS	Rod	7.34	51
<i>Citrus lemon</i>	Fruits	XRD, FTIR, DRS, PL, SEM, TEM	Irregular	75	52
<i>Callistemon viminalis</i>	Flowers	UV-Vis, SEM, TEM, XRD, EDS, Raman	Rod	7.25–7.34	53
<i>Hyphaene thebaica</i>	Fruits	XRD, DRS, FTIR, Zeta potential, Raman, HRSEM, HRTEM, EDS	Rod	7	54
<i>Callistemon viminalis</i>	Flowers	XRD, PL, FTIR, SEM, TEM, EDS	Rod	54	55
<i>Aegle marmelos</i>	Fruits	XRD, SEM, EDS, UV-Vis	-	50	56

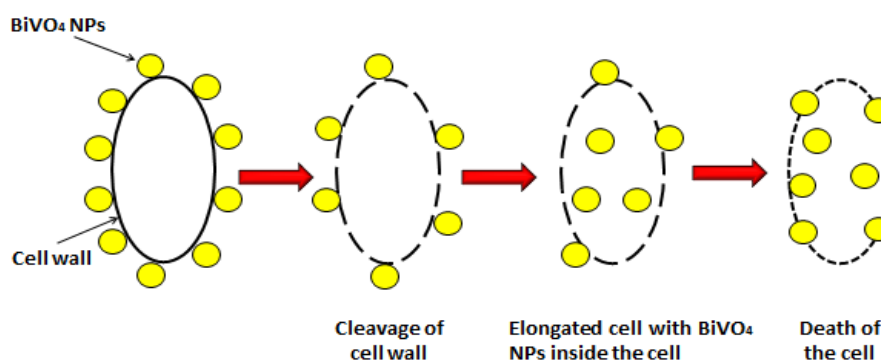


Fig. 6: Possible mechanism of antibacterial activity of BiVO₄ NPs.²⁰

Pramila et al.⁵⁶ described the green synthesis of BiVO₄ NPs using *Aegle marmelos* fruit juice as a fuel and reported their photocatalytic and antimicrobial activity (Fig. 6). They reported that, the excellent photocatalytic degradation of MB dye up to 91% in 160 min under UV irradiation by using BiVO₄ NPs as a photocatalyst. Moreover, as-synthesized BiVO₄ NPs showed good antimicrobial activity against *Staphylococcus aureus*, *Klebsiella pneumonia*, *Enterobacter aerogenes*, *Micrococcus luteus*, *Salmonella typhimurium*, *Proteus vulgaris*, *Salmonella paratyphi-B*, *Candida albicans*, *Malassezia pachydermatis*, *Botrytis cinerea* and *Candida krusei* by using in vitro disc diffusion method. Overall, aforementioned results suggest that as-prepared BiVO₄ NPs via green chemistry approaches will play a vital role as a proper and desirable candidate for photocatalytic and antimicrobial utilization.

CONCLUSION

Green synthesis of BiVO₄ NPs has gained magnificent importance due to its effortless, sustainability, cost effectiveness and eco-benevolent nature. Few plant extracts have been successfully applied for the simple

green synthesis of BiVO₄ NPs. The bioactive natural constituents of the plant extract were found to play double roles like natural reduction and also stabilization of BiVO₄ NPs. UV-Visible spectroscopy, FTIR, XRD, SEM, EDS, TEM, and XPS are the most applied analytical tools for the characterization of BiVO₄ NPs. Moreover, BiVO₄ NPs proved to be advantageous to the sector of catalysis, agriculture, defense, sensing, biomedicine, fuel, water purification with incredible future implication. Further investigation needs to depict the exact mechanism behind the green synthesis of BiVO₄ NPs, because no any reports are existed in literature and only few plants have been reported for green synthesis of BiVO₄ NPs. Thusly, further research needs to highlight the lucid mechanism behind the green synthesis of BiVO₄ NPs. Also, eco-friendly synthesis of BiVO₄ NPs using plants is an area that remains largely unexplored. Therefore, maximum numbers of medicinal plants are needed to use for the synthesis of BiVO₄ NPs without jeopardizing the existing plant diversity. Accordingly, several ways remain for the exploration of novel green preparatory protocols based on eco-friendly synthesis.

REFERENCES

1. Gawande MB, Goswami A, Felpin FX, Asefa T, Huang X, Silva R, Zou X, Zboril R, Varma RS. Cu and Cu-based nanoparticles: synthesis and applications in catalysis. *Chem. Rev.* 2016;116(6):3722-811.
2. Matussin S, Harunsani MH, Tan AL, Khan MM. Plant-extract-mediated SnO₂ nanoparticles: Synthesis and applications. *ACS Sustain. Chem. Eng.* 2020;8:3040-3054.
3. Ghotekar S. A review on plant extract mediated biogenic synthesis of CdO nanoparticles and their recent applications. *Asian J. Green Chem.* 2019;3:187-200.
4. Tarannum N, Gautam YK. Facile green synthesis and applications of silver nanoparticles: a state-of-the-art review. *RSC Adv.* 2019;9:34926-34948.
5. Pagar T, Ghotekar S, Pagar K, Pansambal S, Oza R. A review on bio-synthesized Co₃O₄ nanoparticles using plant extracts and their diverse applications. *J. Chem. Rev.* 2019;1:260-270.
6. Nikam A, Pagar T, Ghotekar S, Pagar K, Pansambal S. A review on plant extract mediated green synthesis of zirconia nanoparticles and applications. *J. Chem. Rev.* 2019;1:154-163.
7. Ghotekar S. Plant extract mediated biosynthesis of Al₂O₃ nanoparticles-a review on plant parts involved, characterization and applications. *Nanochem. Res.* 2019;4:163-169.
8. Oza G, Reyes-Calderón A, Mewada A, Arriaga LG, Cabrera GB, Luna DE, Iqbal HM, Sharon M, Sharma A. Plant-based metal and metal alloy nanoparticle synthesis: a comprehensive mechanistic approach. *J. Mater. Sci.* 2020;55:1309-1330.
9. Ghotekar S, Pansambal S, Pawar SP, Pagar T, Oza R, Bangale S. Biological activities of biogenically synthesized fluorescent silver nanoparticles using *Acanthospermum hispidum* leaves extract. *SN Appl. Sci.* 2019;1:1342.
10. Korde P, Ghotekar S, Pagar T, Pansambal S, Oza R, Mane D. Plant extract assisted eco-benevolent synthesis of selenium nanoparticles: A review on plant parts involved, characterization and their recent applications. *J. Chem. Rev.* 2020;2:157-168.
11. Ghosh Chaudhuri R, Paria S. Core/shell nanoparticles: classes, properties, synthesis mechanisms, characterization, and applications. *Chem. Rev.* 2012;112:2373-2433.
12. Pagar K, Ghotekar S, Pagar T, Nikam A, Pansambal S, Oza R, Sanap D, Dabhane H. Antifungal activity of biosynthesized CuO nanoparticles using leaves extract of *Moringa oleifera* and their characterizations. *Asian J. Nanosci. Mater.* 2020;3:15-23.
13. Kamble DR, Bangale SV, Ghotekar SK, Bamane SR. Efficient synthesis of CeVO₄ nanoparticles using combustion route and their antibacterial activity. *J. Nanostruct.* 2018;8:144-151.
14. Ishak NM, Kamarudin SK, Timmiati SN. Green synthesis of metal and metal oxide nanoparticles via plant extracts: An overview. *Mate. Res. Exp.* 2019;6:112004.
15. Ghotekar S, Pansambal S, Pagar K, Pardeshi O, Oza R. Synthesis of CeVO₄ nanoparticles using sol-gel auto combustion method and their antifungal activity. *Nanochem. Res.* 2018;3:189-196.
16. Savale A, Ghotekar S, Pansambal S, Pardeshi O. Green synthesis of fluorescent CdO nanoparticles using *Leucaena leucocephala* L. extract and their biological activities. *J. Bacteriol. Mycol.* 2017;5:00148.
17. Syedmoradi L, Daneshpour M, Alvandipour M, Gomez FA, Hajghassem H, Omidfar K. Point of care testing: The impact of nanotechnology. *Biosens. Bioelec.* 2017;87:373-387.
18. Ghotekar S, Pagar T, Pansambal S, Oza R. A Review on green synthesis of sulfur nanoparticles via plant extract,

- characterization and its applications. *Adv. J. Chem. B* 2020;2:128-143
19. Ghotekar S, Savale A, Pansambal S. Phytofabrication of fluorescent silver nanoparticles from *Leucaena leucocephala* L. leaves and their biological activities. *J. Water Environ. Nanotechnol.* 2018;3:95-105.
20. Pagar T, Ghotekar S, Pansambal S, Oza R, Marasini BP. Facile plant extract mediated eco-benevolent synthesis and recent applications of CaO-NPs: A state-of-the-art review. *J. Chem. Rev.* 2020;2:201-210.
21. Ghotekar S, Dabhane H, Pansambal S, Oza R, Tambade P, Medhane V. A review on biomimetic synthesis of Ag₂O nanoparticles using plant extract, characterization and its recent applications. *Adv. J. Chem. B* 2020;2:102-111.
22. Pansambal S, Deshmukh K, Savale A, Ghotekar S, Pardeshi O, Jain G, Aher Y, Pore D. Phytosynthesis and biological activities of fluorescent CuO nanoparticles using *Acanthospermum hispidum* L. extract. *J. Nanostruct.* 2017;7:165-174.
23. Pansambal S, Ghotekar S, Shewale S, Deshmukh K, Barde N, Bardapurkar P. Efficient synthesis of magnetically separable CoFe₂O₄@SiO₂ nanoparticles and potent catalytic applications for the synthesis of 5-aryl-1, 2, 4-triazolidine-3-thione derivatives. *J. Water Environ. Nanotechnol.* 2019;4:174-186.
24. Bangale S, Ghotekar S. Bio-fabrication of silver nanoparticles using *Rosa Chinensis* L. extract for antibacterial activities. *Int. J. Nano Dimens.* 2019;10:217-224.
25. Rajeshkumar S, Naik P. Synthesis and biomedical applications of cerium oxide nanoparticles: A review. *Biotechnol. Rep.* 2018;17:1-5.
26. Zhang L, Chen D, Jiao X. Monoclinic structured BiVO₄ nanosheets: hydrothermal preparation, formation mechanism, and coloristic and photocatalytic properties. *J. Phys. Chem. B* 2006;110:2668-2673.
27. Malathi A, Madhavan J, Ashokkumar M, Arunachalam P. A review on BiVO₄ photocatalyst: activity enhancement methods for solar photocatalytic applications. *Appl. Catal. A* 2018;555:47-74.
28. Hu Y, Fan J, Pu C, Li H, Liu E, Hu X. Facile synthesis of double cone-shaped Ag₄V₂O₇/BiVO₄ nanocomposites with enhanced visible light photocatalytic activity for environmental purification. *J. Photochem. Photobiol. A* 2017;337:172-183.
29. Lv D, Zhang D, Pu X, Kong D, Lu Z, Shao X, Ma H, Dou J. One-pot combustion synthesis of BiVO₄/BiOCl composites with enhanced visible-light photocatalytic properties. *Separat. Pur. Technol.* 2017;174:97-103.
30. He H, Berglund SP, Rettie AJ, Chemelewski WD, Xiao P, Zhang Y, Mullins CB. Synthesis of BiVO₄ nanoflake array films for photoelectrochemical water oxidation. *J. Mater. Chem. A* 2014;2:9371-9379.
31. Timmaji HK, Chanmanee WD, De Tacconi NR, Rajeshwar K. Solution combustion synthesis of BiVO₄ nanoparticles: Effect of combustion precursors on the photocatalytic activity. *J. Adv. Oxid. Technol.* 2011;14:93-105.
32. Nguyen DT, Hong SS. Synthesis of BiVO₄ nanoparticles using microwave process and their photocatalytic activity under visible light irradiation. *J. Nanosci. Nanotechnol.* 2017;17:2690-2694.
33. Karunakaran C, Kalaivani S, Vinayagamorthy P, Dash S. Electrical, optical and visible light-photocatalytic properties of monoclinic BiVO₄ nanoparticles synthesized hydrothermally at different pH. *Mater. Sci. Semiconduct. Process.* 2014;21:122-131.
34. Ke D, Peng T, Ma L, Cai P, Jiang P. Photocatalytic water splitting for O₂ production under visible-light irradiation on BiVO₄ nanoparticles in different sacrificial reagent solutions. *Appl. Catal. A* 2008;350:111-117.
35. Ren L, Jin L, Wang JB, Yang F, Qiu MQ, Yu Y. Template-free synthesis of BiVO₄ nanostructures: I. Nanotubes with hexagonal cross sections by oriented attachment and their photocatalytic property for water splitting under visible light. *Nanotechnol.* 2009;20:115603.
36. Castillo NC, Heel A, Graule T, Pulgarin C. Flame-assisted synthesis of nanoscale, amorphous and crystalline, spherical BiVO₄ with visible-light photocatalytic activity. *Appl. Catal. B* 2010;95:335-347.
37. Sivakumar V, Suresh R, Giribabu K, Narayanan V. BiVO₄ nanoparticles: Preparation, characterization and photocatalytic activity. *Cogent Chem.* 2015;1:1074647.
38. Eda SI, Fujishima M, Tada H. Low temperature-synthesis of BiVO₄ nanorods using polyethylene glycol as a soft template and the visible-light-activity for copper acetylacetonate decomposition. *Appl. Catal. B* 2012;125:288-293.
39. Shang M, Wang W, Zhou L, Sun S, Yin W. Nanosized BiVO₄ with high visible-light-induced photocatalytic activity: ultrasonic-assisted synthesis and protective effect of surfactant. *J. Hazard. Mater.* 2009;172:338-344.
40. Venkatesan R, Velumani S, Kassiba A. Mechanochemical synthesis of nanostructured BiVO₄ and investigations of related features. *Mater. Chem. Phys.* 2012;135:842-848.
41. García-Pérez UM, Sepúlveda-Guzmán S, Martínez-De La Cruz A. Nanostructured BiVO₄ photocatalysts synthesized via a polymer-assisted coprecipitation method and their photocatalytic properties under visible-light irradiation. *Solid State Sci.* 2012;14:293-298.
42. Luo Q, Zhang L, Chen X, Tan OK, Leong KC. Mechanochemically synthesized m-BiVO₄ nanoparticles for visible light photocatalysis. *RSC Adv.* 2016;6:15796-15802.
43. Liu W, Wang X, Cao L, Su G, Zhang L, Wang Y. Microemulsion synthesis and photocatalytic activity of visible light-active BiVO₄ nanoparticles. *Sci. China Chem.* 2011;54:724-729.
44. Venkatesan R, Velumani S, Tabellout M, Errien N, Kassiba A. Dielectric behavior, conduction and EPR centres in BiVO₄ nanoparticles. *J. Phys. Chem. Solids* 2013;74:1695-1702.
45. Li C, Pang G, Sun S, Feng S. Phase transition of BiVO₄ nanoparticles in molten salt and the enhancement of visible-light photocatalytic activity. *J. Nanopart. Res.* 2010;12:3069-3075.
46. Ma L, Li WH, Luo JH. Solvothermal synthesis and characterization of well-dispersed monoclinic olive-like BiVO₄ aggregates. *Mater. Lett.* 2013;102:65-67.
47. Zhou L, Wang W, Zhang L, Xu H, Zhu W. Single-crystalline BiVO₄ microtubes with square cross-sections: microstructure, growth mechanism, and photocatalytic property. *J. Phys. Chem. C* 2007;111:13659-13664.
48. Wang M, Liu Q, Luan HY. Preparation, Characterization and photocatalytic property of BiVO₄ photocatalyst by Sol-Gel method. *Appl. Mech. Mater.* 2011;99:1307-1311.
49. Liu W, Cao L, Su G, Liu H, Wang X, Zhang L. Ultrasound assisted synthesis of monoclinic structured spindle BiVO₄ particles with hollow structure and its photocatalytic property. *Ultrason. Sonochem.* 2010;17:669-674.
50. Galembeck A, Alves OL. Bismuth vanadate synthesis by metallo-organic decomposition: thermal decomposition study and particle size control. *J. Mater. Sci.* 2002;37:1923-1927.

51. Mohamed HE, Sone BT, Khamlich S, Coetsee-Hugo E, Swart HC, Thema T, Sbiaa R, Dhlamini MS. Biosynthesis of BiVO₄ nanorods using *Callistemon viminalis* extracts: Photocatalytic degradation of methylene blue. Mater. Today 2020; in press.
52. Manjunatha AS, Pavithra NS, Marappa S, Prashanth SA, Nagaraju G. Green synthesis of flower-like BiVO₄ nanoparticles by solution combustion method using lemon (*Citrus Limon*) juice as a fuel: Photocatalytic and electrochemical study. Chem. Select 2018;3:13456-13463.
53. Mohamed HE, Sone BT, Dhlamini MS, Maaza M. Bio-synthesis of BiVO₄ nanorods using extracts of *Callistemon viminalis*. MRS Adv. 2018;3:2479-2486.
54. Mohamed HE, Afridi S, Khalil AT, Zohra T, Alam MM, Ikram A, Shinwari ZK, Maaza M. Phytosynthesis of BiVO₄ nanorods using *Hyphaene thebaica* for diverse biomedical applications. AMB Exp. 2019;9:1-4.
55. Mohamed HE, Sone BT, Fuku XG, Dhlamini MS, Maaza M. Green synthesis of BiVO₄ nanorods via aqueous extracts of *Callistemon viminalis*. AIP Conf. Proc. 2018;1962:040004.
56. Pramila S, Nagaraju G, Mallikarjunaswamy C, Latha KC, Chandan S, Ramu R, Rashmi V, Lakshmi Ranganatha V. Green Synthesis of BiVO₄ nanoparticles by microwave method using *Aegle marmelos* juice as a fuel: Photocatalytic and antimicrobial study. Anal. Chem. Lett. 2020;10:298-306.
57. Guo M, He Q, Wang A, Wang W, Fu Z. A novel, simple and green way to fabricate BiVO₄ with excellent photocatalytic activity and its methylene blue decomposition mechanism. Cryst. 2016;6:81.

How to cite this article: Ghotekar S, Pagar K, Pansambal S, Murthy HCA, Oza R. A review on eco-friendly synthesis of BiVO₄ nanoparticle and its eclectic applications. Adv. J. Sci. Eng. 2020;1(4):106-112.

DOI: <http://doi.org/10.22034/AJSE2014106>

URL: <http://ajscieng.com/index.php/ajse/article/view/ajse2014106>